

(b) Explain the following terms : **14**

- (i) Photo-fries reaction of anilides
- (ii) Singlet molecular oxygen reaction
- (iii) Dimerization reaction of carbonyl compound
- (iv) Photodegradation of polymer.

4. (a) Discuss the Paterno-Buchi reaction with mechanism and its stereochemical consequences. **8**
- (b) Give mechanism and synthetic applications of Barton reaction. **8**
- (c) Write the products of bromination of 1-butene with NBS and indicate the major product. **4**

Unit III

5. (a) Under photochemical conditions, will ring closure of (2E, 4Z, 6Z, 8E)-decatetraene be conrotatory or disrotatory? Will the product have the cis or the trans configuration? **6**

M-DD-297

4

No. of Printed Pages : 07

Roll No.

DD-297

M. Sc. EXAMINATION, May 2017

(Fourth Semester)

(Main & Re-appear)

CH-614-B

CHEMISTRY

Organic Chemistry Special-IV

(Photochemistry and Pericyclic Reactions)

Time : 3 Hours]

[Maximum Marks : 100

Before answering the question-paper candidates should ensure that they have been supplied to correct and complete question-paper. No complaint, in this regard, will be entertained after the examination.

Note : Attempt *Five* questions in all, selecting at least *one* question from each Unit. All questions carry equal marks.

(3-22/3) M-DD-297

P.T.O.

Unit I

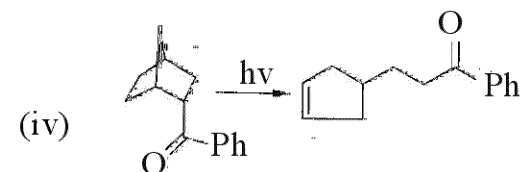
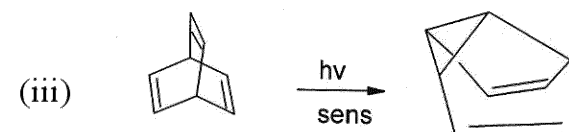
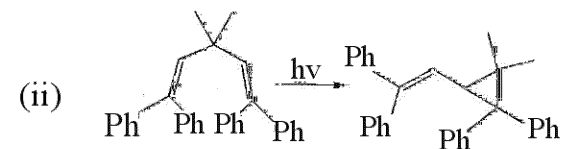
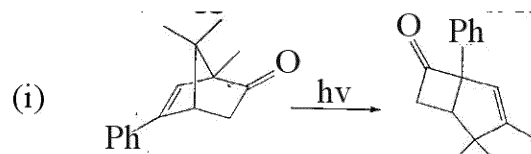
1. Write brief notes on the following : 10

- (a) Explain electronic excitation and their selection rules. Discuss fate of excited molecule (Jablonskii diagram) and term involved.
- (b) Discuss the photochemical rearrangement of 1, 4 and 1, 5-dienes with examples. 10

2. (a) Explain the following terms : 9

- (i) Quantum yield and Actinometry
- (ii) Importance of photosensitiser in photochemistry
- (iii) Norrish Type-II process with example

(b) Suggest mechanism for the following reactions : 8



(c) Explain, the photochemical isomerisation (geometrical) of alkenes. 3

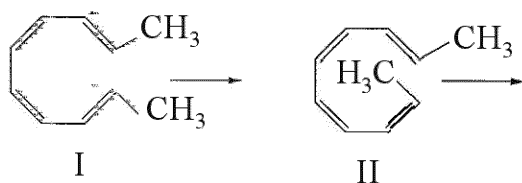
Unit II

3. (a) Irradiation of o-xylene gives a mixture of m- and p-xylene, proposed a mechanism for this transformation. 6

- (b) Explain, why maleic anhydride reacts rapidly with 1,3-butadiene but does not react at all with ethene under thermal conditions. **4**
- (c) With the help of correlation diagram and PMO approach prove that Diels-Alder reaction is thermally allowed. **6**
- (d) Discuss [1, 3] sigmatropic rearrangement with mechanism. **4**
6. (a) 5-methyl-1,3-cyclopentadiene rearranges to give a mixture of 5-methyl-1,3-cyclopentadiene, 1-methyl-1,3-cyclopentadiene, and 2-methyl-1,3-cyclopentadiene. Show how these products are formed. **4**
- (b) Write explanatory notes on the following : **6**
- (i) Claisen rearrangement
- (ii) [1, 5] sigmatropic rearrangement.

- (c) Give the product formed when each of the following compounds undergoes an electrocyclic reaction : **6**

- (i) Under thermal conditions
(ii) Under photochemical conditions.



- (d) With the help of FMO methods show that the [2+2] cycloaddition reaction between alkene and ketene is thermal allowed reaction. **4**

Unit IV

7. Write short notes on the following : **20**

- (a) Anomeric effect and double Anomeric effect
(b) Ring inversion and pyramidal inversion

- (c) Conformational analysis of medium ring compounds
(d) 2-alkyl ketone and 3-alkyl ketone effects in cyclohexanone.

8. (a) With suitable examples discuss the stereochemistry of nitrogen containing compounds, strain and their consequences in small ring heterocycles. **9**
(b) Write an explanatory note on conformation of cyclodecane. **8**
(c) Explain, trans annular reactions with minimum two examples. **3**